Three New Sesquiterpenes from *Croton arboreous*

A. Berenice Aguilar-Guadarrama and María Yolanda Rios*

Centro de Investigaciones Químicas de la Universidad Autónoma del Estado de Morelos, Col. Chamilpa, 62210 Cuernavaca, Morelos, México

Received November 6, 2003

Three new sesquiterpenes, 5α , 7α , 10β H-3-patchoulen-2-one (**1**), 5α , $7\alpha 10\beta$ H-4(14)-patchoulen-2 α -ol (**2**), and 9α , 10β -dihydroxy- 2β , 4β -peroxy- 1α , 5β , 7α H-guaiane (**3**), were isolated from the aerial parts of *Croton arboreous* along with 14 known compounds. The structures of these compounds were determined on the bases of their spectroscopic data (IR, UV, OR, 1D and 2D NMR, and MS). The anti-inflammatory activity against ear edema in mice produced by 12-*O*-tetradecanoylphorbol-13-acetate (TPA) was evaluated for all the pure compounds and showed that compounds **4**–**7** are active.

The aerial parts of *Croton arboreous* (Euphorbiaceae), common name "cascarillo", are the source of a popular beverage used in Tabasco and Chiapas, Mexico, as an auxiliary anti-inflammatory agent in the treatment of respiratory ailments. The *n*-hexane, acetone, and methanol extracts of this plant were analyzed in the anti-inflammatory assay against ear edema in mice produced by TPA (0.5 mg/ear),^{1,2} and the *n*-hexane and acetone extracts were active. The *n*-hexane and acetone extracts were analyzed on TLC, and on the basis of their similar chemical composition and anti-inflammatory activity these extracts were combined. Purification of the nonactive fractions yielded 10 known natural products and three new sesquiterpenes, 5α , 7α , $10\beta H$ -3-patchoulen-2-one (1), 5α , 7α , $10\beta H$ -4(14)patchoulen- 2α -ol (**2**), and 9α , 10β -dihydroxy- 2β , 4β -peroxy- $1\alpha, 5\beta, 7\alpha H$ -guaiane (3), whose structure elucidation is described herein. Fractionation of the active residue on CC afforded three active groups of fractions, whose further purification yielded four anti-inflammatory compounds, 5α , 10β -4(15)-eudesmen- 1β , 6β -diol (4), ³ spathulenol (5), ⁴ 5α , 10β -3-eudesmen- 1β , 6α -diol (6), 5 and junceic acid (7). 6β

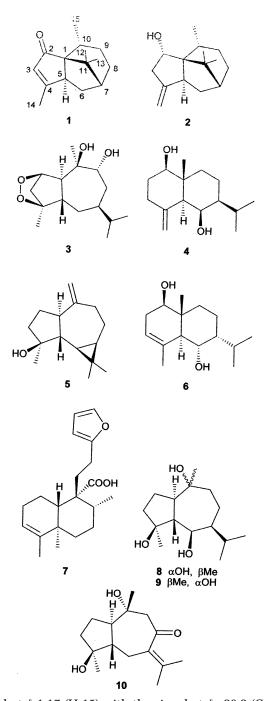
Compound 1 was isolated as a yellowish oil. The HREIMS data indicated a molecular formula C15H22O on the basis of the m/z 218.3383 and five unsaturation degrees, two of them due to the presence of a α,β -unsaturated ketone in accord with the absorptions at 1669 and 1457 cm^{-1} in the IR spectrum. The ¹³C NMR and DEPT spectra of 1 showed 15 carbon resonances, corresponding to four CH₃, three CH₂, four CH, and four C, establishing a patchoulene skeleton for this natural product. This deduction was supported by the ¹H NMR and HMQC spectra, where 1 showed signals of four methyl groups, two singlets at δ 1.14 $(\delta_{\rm C} 27.1, \text{ C-13})$ and 1.01 $(\delta_{\rm C} 28.0, \text{ C-12})$ and two doublets at δ 1.29 (d, J = 7.1 Hz, $\delta_{\rm C}$ 15.3, C-15) and δ 1.98 (d, J =1.2 Hz, $\delta_{\rm C}$ 25.5, C-14), one vinyl hydrogen at δ 5.65 (q, J =1.2 Hz, $\delta_{\rm C}$ 125.6, C-3), and the bridged hydrogen H-7 at δ 1.99–1.91 (m, $\delta_{\rm C}$ 58.3). Long-range ¹H–¹³C couplings observed in the HMBC spectrum supported the structure shown for 1: the coupling between δ 2.33 (H-5) and the signals at $\delta_{\rm C}$ 65.6 (C-1), 165.3 (C-4), 125.6 (C-3), 25.5 (C-14), and 58.3 (C-7) revealed that the α,β -unsaturated ketone is located in the A ring; that between δ 1.14 (H-13) and 1.01 (H-12) and the signals at $\delta_{\rm C}$ 42.5 (C-11) and 58.3 (C-7) confirmed the connectivity of the five-membered B ring, and that between δ 1.29 (H-15) and the signals at $\delta_{\rm C}$ 65.6 (C-1), 44.0 (C-10), and 39.7 (C-9) revealed the connectivity of the six-membered C ring. The configuration of **1** was assigned by the observation of three important correlations in the NOESY spectrum (Figure 1). The signal at δ 2.33 (H-5) showed correlation to H-6a (δ 1.75), while the signal at δ 1.94 (H-6b) showed correlation to H-12 (δ 1.01). Additionally the methyl group C-13 showed strong correlations to protons H-10 (δ 1.69) and H-8b (δ 1.49). Thus, the structure of **1** was established as 5α , 7α , 10β H-3-patchoulen-2-one.

Compound 2 was an alcohol in accord with its absorption at 3408 cm⁻¹ in the IR spectrum and the signal at δ 3.84 in ¹H NMR. The spectroscopic data in the ¹H, ¹³C NMR and ¹³C DEPT spectra were very similar with those of compound 1. However, 2 showed only three signals for methyl groups [singlets at δ 1.01 (H-13) and 0.96 (H-12) and a doublet at δ 1.14 (H-15)] and one AB system at δ 4.59 and 4.55, which was assigned to an exocyclic methylene (H-14). In the COSY spectrum, both signals for H-14 showed correlation with the signal at δ 2.14–2.05 (m, H-3a) and also with the signal for H-3b at δ 2.69 and between the latter two with the signal at δ 3.84, establishing that the alcohol was located at C-2. The set of correlations shown in the NOESY spectrum for 1 were also observed for **2**, in addition to the correlation between H-10 (δ 1.61) and H-2 (δ 3.84), establishing an α orientation of the hydroxyl group on C-2 and the structure 5α , 7α , 10β H-4(14)patchoulen- 2α -ol for compound **2**.

Compound 3 was a yellowish oil with molecular formula $C_{15}H_{26}O_4$ (M⁺ – O₂, *m*/*z* 238.1933), indicating three unsaturation degrees. The ¹³C NMR spectrum of **3** showed 15 carbon resonances, corresponding to four CH₃, three CH₂, six CH, and two C in accord with the DEPT spectrum. The two quaternary carbons of **3** showed binding to oxygen, in accord with their chemical shifts at $\delta_{\rm C}$ 87.4 (s) and 77.4 (s). This characteristic and the multiplicity of the 13 other carbons established a totally saturated guaiane skeleton for this natural product, which justified two unsaturation degrees. The two quaternary carbons corresponded to C-4 and C-10 in accord with the chemical shift for the methyl groups C-14 and C-15 at δ 1.49 (s, δ_C 21.5) and 1.17 (s, δ_C 32.1), respectively, in the ¹H, ¹³C NMR and HMQC spectra. The presence of two additional tertiary carbons joined to oxygen was established by the signals at δ 3.91 and 3.80. The first one corresponds to H-9 in accord with a correlation of this signal with the signal at δ 1.68 (H-8a) in the COSY spectrum and with the long-range ${}^{1}H^{-13}C$ coupling of the

10.1021/np030485f CCC: \$27.50 © 2004 American Chemical Society and American Society of Pharmacognosy Published on Web 05/01/2004

^{*} To whom correspondense should be addressed. Fax: +52 (777) 329 79 97. E-mail: myolanda@uaem.mx.



signal at δ 1.17 (H-15) with the signal at $\delta_{\rm C}$ 86.8 (C-9) in the HMBC spectrum. The second one corresponded to H-2 in accord with the correlation of the signal with the signals at δ 0.98 (H-3a) and 1.41 (H-3b) in the COSY spectrum. The downfield shift observed in the $^1\mathrm{H}$ and $^{1\bar{3}}\mathrm{C}$ NMR spectra for H-14 ($\Delta\delta$ 1.49–1.20 = 0.29) and C-4 ($\Delta\delta$ 87.4– 81.2 = 7.2) with respect to teuclatriol (8)⁷ and $\Delta\delta$ 1.49– 1.26 = 0.23 and $\Delta \delta 87.4 - 81.1 = 7.3$ with respect to 10epiteuclatriol (9)⁷ established a peroxide function on C_4 -C₂, justifying the third unsaturation in this natural product. The other two oxygenated functions correspond to hydroxyl groups, in accord with the absorption at 3398 cm⁻¹ in the IR spectrum. The junction of the guaiane rings was trans, in accord with a $J_{H1-H5} = 12.8$ Hz.^{7,8} The relative configuration at C-10 was established by comparison of the chemical shift due to C-15 in 3 with those of zedoarondiol (**10**, δ_{C15} **20.6**), a guaiane whose 10 α -hydroxy configuration has been determined by X-ray analysis,⁹ and the epimeric compounds at C-10 teuclatriol (8, 10 α -hydroxy, δ_{C15} 22.2)

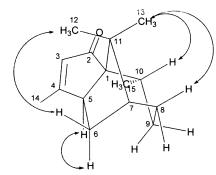


Figure 1. Spatial correlations observed from NOESY spectra of 1 and 2.

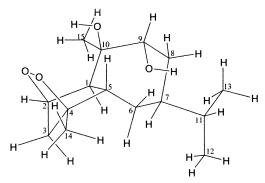


Figure 2. Energy-minimized molecular model of 3.

and epiteuclatriol (9, 10 β -hydroxy, δ_{C15} 29.9).⁷ Compound **3** showed a chemical shift for C-15 similar to **9** at δ 32.1, establishing a β orientation for the hydroxyl group and an α orientation for C-15. A trans and syn relationship between H-1 and H-2 should have a $J_{H1\alpha-H2\beta} = 9.0$ Hz and $J_{\rm H1\alpha-H2\alpha}$ = 5.0 Hz, respectively.8 Compound 3 showed a $J_{\text{H1}\alpha-\text{H2}\alpha} = 4.8$ Hz, corresponding to a *syn* relationship, establishing that the C_4-C_2 peroxide was β . Additionally observed coupling constants were $J_{H5\beta-H6\alpha} = 9.2$ Hz and $J_{\rm H5\beta-H6\beta} = 4.0$ Hz, $J_{\rm H6\alpha-H7\alpha} = 5.2$ Hz, and $J_{\rm H8\alpha-H9\beta} = 5.2$ Hz, which could be explained in accord with the H-H dihedral angles of 147°, 27°, 47°, and 48°, respectively, obtained from the energy-minimized molecular model calculation (Figure 2). Thus, the structure of 3 was established as 9α , 10β -dihydroxy- 2β , 4β -peroxy- 1α , 5β , 7α Hguaiane and was confirmed by the HMBC spectrum, where 3 showed long-range ${}^{1}\text{H}{-}{}^{13}\text{C}$ couplings between δ 0.99 (H-13) and 0.98 (H-12) and the signals at $\delta_{\rm C}$ 47.5 (C-7) and 28.5 (C-11); between δ 1.49 (H-14) and the signals at $\delta_{\rm C}$ 87.4 (C-4), 30.2 (C-3), and 61.8 (C-5); between δ 1.17 (H-15) and the signals at $\delta_{\rm C}$ 41.6 (C-1) and 86.8 (C-9); between δ 1.30 (H-6) and 61.8 (C-5); and between δ 1.24 (H-8) and 86.8 (C-9).

Experimental Section

General Experimental Procedures. The *n*-hexaneacetone extract from *C. arboreous* was fractionated using open CC (Merck Kiesel-gel 60 and Supelclean SPE LC-SI 6 mL tubes) and TLC (ALUGRAM SIL G/UV₂₅₄ silica gel plates), using mixtures of *n*-hexane-acetone as eluent. In the TLC analysis, the compounds were visualized by UV light and spraying with a 1% solution of $(NH_4)_4Ce(SO_4)$ in 2 N H₂SO₄; UV spectra were recorded on a Hewlett-Packard 8453 spectrophotometer in CHCl₃. IR spectra were recorded with Bruker Vector 22 IR instrument in CHCl₃ solution; ¹H and ¹³C NMR spectra as well as 2D NMR experiments were recorded in CDCl₃ on Varian Unity 400 and Varian-Gemini 300 spectrometers, and the chemical shifts were expressed in parts per million (δ) relative to TMS as internal standard. Mass spectra were measured on a JEOL JMS-AX 505 HA mass spectrometer. Electron impact mass spectra were obtained at 70 eV ionization energy.

Plant Material. The aerial parts of C. arboreous (6 m tree with red resin) were collected and identified by Biol. Esteban Manuel Martínez on March 13, 2002, at 4 km SE of La Nueva Vida, Camino a Xpujil (18°46'30 N, 89°22'23 W) at 270 m above sea level in Calakmul, Campeche, México. A voucher specimen was deposited at the National Herbarium (MEXU, voucher 27171) at the Instituto de Biología, UNAM, México.

Extraction and Isolation. The air-dried parts of C. arboreous (3.48 kg) were powdered and exhaustively extracted with *n*-hexane, acetone, and MeOH (12 L \times 3 each) to yield 45.8, 38.6, and 148.1 g of residue, respectively. These extracts were evaluated in the anti-inflammatory assay,^{1,2} the *n*-hexane and acetone extracts being active. On the basis of their similar anti-inflammatory activity and chemical composition on TLC, the *n*-hexane and acetone extracts were joined and absorbed on silica gel (85 g) and chromatographed on CC over silica gel 60 (850 g), using a gradient of *n*-hexane-acetone as eluent. The composition of the obtained fractions (500 mL each) was monitored by TLC, and the chromatographically identical fractions were combined, yielding eight groups, G-1 [3.9 g, n-hexane 100%], G-2 [9.8 g, n-hexane-acetone, 95:5], G-3 [8.3 g, *n*-hexane-acetone, 95:5], G-4 [7.3 g, *n*-hexane-acetone, 9:1], G-5 [2.3 g, n-hexane-acetone, 85:15], G-6 [1.7 g, n-hexaneacetone, 8:2], G-7 [2.6 g, n-hexane-acetone, 7:3], and G-8 [3.1 g, n-hexane-acetone, 6:4], which were evaluated in the antiinflammatory assay.^{1,2} G-2, G-3, and G-5 displayed antiinflammatory activity. Each group was further separated using CC over silica gel 60, with a gradient of *n*-hexane-acetone as eluent. Fraction G-2 yielded junceic acid⁶ (7, 236 mg, 0.52%) and β -sitosterol. Fraction G-3 yielded 1β , 10α - 4β , 5α -diepoxy- $7\alpha H$ -germacran-6 β -ol¹⁰ ([α]²⁵_D –28.6° (*c* 0.2, CHCl₃), 96 mg, 0.21%), 1α , 10β - 4β , 5α -diepoxy- 7α *H*-germacran- 6β -ol¹¹ ($[\alpha]^{2\bar{5}_{D}}$ $+152.2^{\circ}$ (c 0.5, CHCl₃), 52 mg, 0.11%), 5 α , 10 β -4(15)-eudesmen- 1β ,6 α -diol³ ([α]²⁵_D -51.6° (*c* 0.1, CHCl₃), 193 mg, 0.42%), and 5α , 10β -4(15)-eudesmen- 1β , 6β -diol³ ([α]²⁵_D +15.7° (*c* 0.1, CHCl₃), **4**, 177 mg, 0.39%). Fraction G-5 yielded spathulenol⁴ ($[\alpha]^{25}_{D}$ +3.7° (c 0.01, CHCl₃), **5**, 259 mg, 0.57%), 5α, 10β-3-eudesmen- 1β , 6α -diol⁵ ([α]²⁵_D +44.0° (*c* 0.52, CHCl₃), **6**, 48 mg, 0.11%), and 9α , 10β -dihydroxy- 2β , 4β -peroxy- 1α , 5β , $7\alpha H$ -guaiane ($[\alpha]^{25}_{D}$ +8.3° (*c* 0.05, CHCl₃), **3**, 39 mg, 0.08%). Additional constituents of the inactive fractions were 5α , 7α , 10β H-3-patchoulen-2-one (1, 47 mg, 0.10%), 5α , 7α , 10β H-4(14)-patchoulen-2\alpha-ol (2, 56 mg, 0.12%), lupenone¹² (286 mg, 0.62%), 4β , 6β -dihydroxy- $1\alpha H, 5\beta H-9$ -guaiene¹³ ([α]²⁵_D +10.5° (*c* 0.60, CHCl₃), 88 mg, 0.19%), teucladiol⁷ ($[\alpha]^{25}_{D}$ +2.1° (*c* 0.43, CHCl₃), 79 mg, 0.17%), quercetin-3,4'-dimethyl ether¹⁴ (266 mg, 0.58%), 2'-O- β -Dglucopyranosyl-5,7,5'-trihydroxy-3,4'-dimethoxyflavone¹⁵ (141 mg, 0.31%), and β -sitosteryl- β -D-glucopyranoside.¹⁶

5 α ,**7** α ,**10** β *H***-3**-**Patchoulen-2-one (1)**: yellowish oil; $[\alpha]^{25}_{D}$ -53.9° (*c* 3.6, CHCl₃); UV (CHCl₃) λ_{max} (log ϵ) 239 (3.68), 205 (3.40) nm; IR (CHCl₃) v_{max} 2946, 2870, 1669 (CO), 1457 (C= C), 1375 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 5.65 (1H, q, J =1.2 Hz, H-3), 2.33 (1H, d, J = 3.6 Hz, H-5), 1.99-1.91 (1H, m, H-7), 1.98 (3H, d, J = 1.2 Hz, H-14), 1.94 (1H, d, J = 11.2 Hz, H-6b), 1.94 (1H, m, H-9a), 1.75 (1H, ddd, J = 11.2, 3.6, 3.6 Hz, H-6a), 1.70 (1H, dd, J = 11.2, 5.2 Hz, H-8a), 1.69 (1H, ddd, J = 8.4, 7.1, 4.8 Hz, H-10), 1.62 (1H, dd, J = 10.4, 5.2 Hz, H-9b), 1.49 (1H, dd, J = 11.2, 4.4 Hz, H-8b), 1.29 (3H, d, J = 7.1 Hz, H-15), 1.14 (3H, s, H-13), 1.01 (3H, s, H-12); ¹³C NMR (CDCl₃, 75 MHz) & 203.8 (C, C-2), 165.3 (C, C-4), 125.6 (CH, C-3), 65.6 (C, C-1), 59.1 (CH, C-5), 58.3 (CH, C-7), 44.8 (CH₂, C-6), 44.0 (CH, C-10), 42.5 (C, C-11), 39.7 (CH₂, C-9), 28.0 (CH₃, C-12), 27.1 (CH₃, C-13), 26.7 (CH₂, C-8), 25.5 (CH₃, C-14), 15.3 (CH₃, C-15); EIMS m/z 218 [M⁺, C₁₅H₂₂O] (100), 203 $[M^+ - CH_3]$ (27), 176 $[M^+ - C_3H_6]$ (43), 175 $[M^+ - C_3H_6]$ - H] (42), 161 $[M^+ - C_3H_6 - CH_3]$ (51), 147 $[M^+ - C_3H_6 - CH_3]$ CH₃ - CH₂] (20), 136 (29), 135 (28), 121 (28), 95 (15), 91 (13), 81 (9), 55 (9), 41 (12); HREIMS m/z 218.3383 (calcd for C15H22O, 218.3390).

5 α ,7 α ,10 β H-4(14)-Patchoulen-2 α -ol (2): colorless oil; $[\alpha]^{25}$ _D -16.4° (c 1.2, CHCl₃); IR (CHCl₃) v_{max} 3408 (OH), 2927, 2862, 1463 (C=C), 1387 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 4.59 (1H, td, J = 3.0, 0.9 Hz, H-14a), 4.55 (1H, dd, J = 3.0, 3.0 Hz, H-14b), 3.84 (1H, dd, J = 9.3, 6.9 Hz, H-2), 2.69 (1H, dd, J = 15.3, 6.9 Hz, H-3b), 2.18 (1H, d, J = 4.2 Hz, H-5), 2.14-2.05 (1H, m, H-3a), 1.90 (1H, m, H-9a), 1.72 (1H, dd, J = 12.3, 3.2)Hz, H-6b), 1.61(2H, m, H-7, H-10), 1.59 (1H, m, H-8a), 1.58 (1H, m, H-6a), 1.42 (1H, m, H-9b), 1.40 (1H, m, H-8b), 1.14 (3H, d, *J* = 7.5 Hz, H-15), 1.01 (3H, s, H-13), 0.96 (3H, s, H-12); ¹³C NMR (CDCl₃, 75 MHz) δ 148.8 (C, C-4), 108.2 (CH₂, C-14), 73.4 (CH, C-2), 60.6 (CH, C-7), 58.3 (C, C-1), 52.5 (CH, C-5), 45.3 (CH, C-10), 42.8 (CH₂, C-6), 41.5 (C, C-11), 40.3 (CH₂, C-9), 40.2 (CH₂, C-3), 27.0 (CH₃, C-13), 26.6 (CH₂, C-8), 25.6 (CH₃, C-12), 14.7 (CH₃, C-15); EIMS *m*/*z* 220 [M⁺, C₁₅H₂₄O] (26), 202 $[M^+ - H_2O]$ (100), 160 $[M^+ - H_2O - C_3H_6]$ (41), 159 $[M^+ - H_2O - C_3H_6 - H]$ (29), 137 (14), 97 (8); HREIMS m/z220.3553 (calcd for C15H24O, 220.3549).

9 α ,10 β -Dihydroxy-2 β ,4 β -peroxy-1 α ,5 β ,7 α *H*-guaiane (3): yellowish oil; $[\alpha]_D^{25} = 8.4^\circ$ (c 0.5, CHCl₃); IR (CHCl₃) ν_{max} 3398 (OH), 2928, 1377, 1266, 1005, and 978 cm⁻¹; ¹H NMR (CDCl₃, 400 MHz) δ 3.91 (1H, d, J = 5.2 Hz, H-9), 3.80 (1H, ddd, J =4.8, 4.4, 3.2 Hz, H-2), 2.15 (1H, dd, J = 12.8, 4.8 Hz, H-1), 1.98 (1H, ddd, J = 12.8, 9.2, 4.0 Hz, H-5), 1.68 (1H, m, H-8a), 1.66 (1H, dh, J = 6.8, 2.4 Hz, H-11), 1.63 (1H, m, H-7), 1.53 (1H, dd, J = 9.2, 5.2 Hz, H-6b), 1.49 (3H, s, H-14), 1.41 (1H, 1)m, H-3b), 1.30 (1H, m, H-6a), 1.24 (1H, m, H-8b), 1.17 (3H, s, H-15), 0.99 (3H, d, J = 6.8 Hz, H-13), 0.98 (3H, d, J = 6.8 Hz, H-12), 0.98 (1H, m, H-3a); ¹³C NMR (CDCl₃, 100 MHz) δ 87.4 (C, C-4), 86.8 (CH, C-9), 77.4 (C, C-10), 68.7 (CH, C-2), 61.8 (CH, C-5), 47.5 (CH, C-7), 41.6 (CH, C-1), 32.1 (CH₃, C-15), 31.5 (CH2, C-6), 30.2 (CH2, C-3), 28.5 (CH, C-11), 26.6 (CH2, C-8), 21.5 (CH₃, C-14), 21.4 (CH₃, C-12), 21.3 (CH₃, C-13); EIMS m/z 238 [M⁺ - O₂, C₁₅H₂₆O₂] (9), 220 [M⁺ - O₂ - H₂O] (23), 205 [M⁺ - O₂ - H₂O - CH₃] (9), 195 [M⁺ - O₂ - C₃H₇] (50), 181 $[M^+ - O_2 - C_3H_6 - CH_3]$ (39), 177 (38), 162 (45), 152 (55), 140 (37), 123 (81), 109 (83), 94 (100), 81 (53), 43 (67); HREIMS m/z 238.1933 (calcd for C₁₅H₂₆O₂, M⁺ – O₂, 238.3702).

Anti-inflammatory Assay. The anti-inflammatory activity of extracts, fractions, and pure compounds at a dose of 0.5 mg/ ear was evaluated against the ear edema in mice produced by 12-O-tetradecanoylphorbol-13-acetate (TPA), according to the previously described procedure,^{1,2} using indomethacin (61.2% inhibition, 0.5 mg/ear) as positive control. The n-hexane and acetone extracts showed activity (48.6 and 54.1% of inhibition, respectively) and were combined for fractionation. Fractions G-2, G-3, and G-5 were active (32.9, 22.7, and 29.3% inhibition), which after purification afforded the active natural products 5α , 10β -4(15)-eudesmen- 1β , 6β -diol (4), 3 spathulenol (5),⁴ 5α , 10β -3-eudesmen- 1β , 6α -diol (6),⁵ and junceic acid (7),⁶ which showed 29.9, 31.5, 28.0, and 38.4% inhibition, respectively.

Acknowledgment. We are indebted to E. Salazar Leyva and R. Patiño Maya for technical assistance. This work was financially supported by CONACyT (Project No. 40405).

References and Notes

- (1) Rao, T. S.; Currie, J. L.; Shaffer, A. F.; Isakson, P. C. Inflammation 1993, 17, 723-741.
- García-Argáez, A. N.; Ramírez, A. T.; Parra, H.; Velázquez, G.; Martínez-Vázquez, M. *Planta Med.* **2000**, *66*, 279–281. (2)
- Kamel A. J. Nat. Prod. 1995, 58, 428-431.
- (d) Inagaki, F.; Abe, A. J. Chem. Soc., Perkin Trans 21985, 1773–1778.
 (d) Inagaki, F.; Abe, A. J. Chem. Soc., Perkin Trans 21985, 1773–1778.
 (f) Mahmoud, A. Phytochemistry 1997, 45, 1633–1638.
 (g) (a) Henderson, M. S.; Murray, R. D. H.; McCrindle, R.; McMaster, D. Can. J. Chem. 1973, 51, 1322–1331. (b) Osakawa, Y.; Toyota, M.; Can. J. Chem. 1973, 51, 1322–1331. (b) Osakawa, Y.; Toyota, M.; Can. J. Chem. 1973, 51, 1322–1331. (b) Osakawa, Y.; Toyota, M.; Can. J. Chem. 1973, 51, 1322–1331. (b) Osakawa, Y.; Toyota, M.; Can. J. Chem. 1973, 51, 1322–1331. (b) Osakawa, Y.; Toyota, M.; Can. J. Chem. 1973, 51, 1322–1331. (c) Osakawa, Y.; Toyota, M.; Can. J. Chem. 1973, 51, 1322–1331. (b) Osakawa, Y.; Toyota, M.; Can. J. Chem. 1973, 51, 1322–1331. (c) Osakawa, Y.; Toyota, Y.; Can. J. Chem. 19 Veda, A. Phytochemistry 1990, 29, 2165–2167. (c) Helvani, C. S.; Catalán, C. A. N.; Hernández, L. R.; Burgueño-Tapia, E.; Joseph-Nathan, P. Magn. Reson. Chem. **1998**, 36, 947–950.
- (7) Bruno, M.; De la Torre, M. C.; Rodríguez, B.; Omar, A. A. Phytochem-istry 1993, 34, 245-247.
- (8) Kitajima, J.; Suzuki, N.; Tanaka, Y. Chem. Pharm. Bull. 1998, 46, 1743-1747.
- (9) Kuroyanagi, M.; Ueno, A.; Ujiie, K.; Sato, S. Chem. Pharm. Bull. 1987, 35, 53-57
- (10)Sanz, J. F.; Marco, J. A. Phytochemistry 1991, 30, 2788-2790.
- Al Yuosuf, M. H.; Bashir, A. K.; Crabb, T. A.; Blunden, G.; Yang, M. *Biochem. Syst. Ecol.* **1999**, *27*, 107–109. (11)

- Wenkert, E.; Baddeley, G. V.; Burfitt, I. R.; Moreno, L. N. Org. Magn. Reson. **1978**, 11, 337–343.
 Bauer, R. F. X.; Khan, I. A.; Lotter, H.; Wagner H.; Wray, V. Helv. Chim. Acta **1985**, 68, 2355–2358.
 Kupcham, S. M.; Bauerschmidt, E. Phytochemistry **1971**, 10, 664– 666.

- (15) Li, R. Z.; Niambai, F.; Mabry, T. J. Phytochemistry 1987, 26, 2831– 2833.
- (16) Voutquenne, L.; Lavaud, C.; Massiot, G.; Sevenet, T.; Hadi, H. A. *Phytochemistry* 1999, *50*, 63–69.

NP030485F